Search for Life

• 1st find <u>water</u>

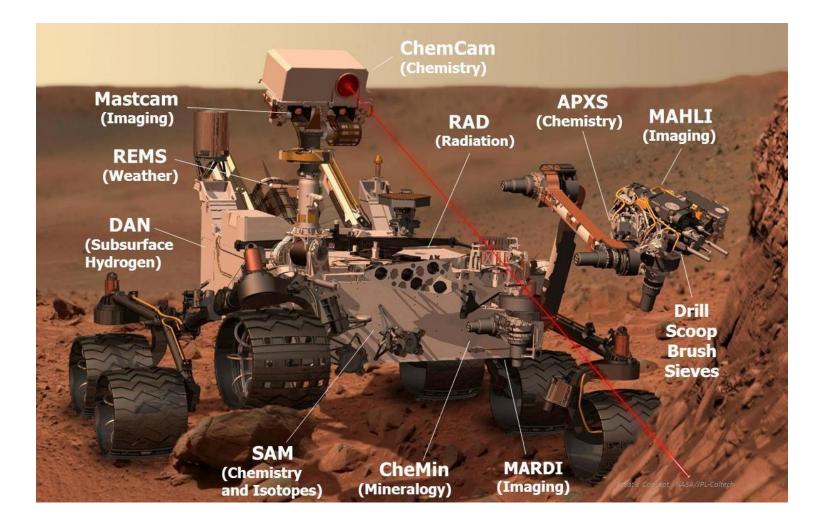
Phoenix found water in 2008



Search for life

- 2nd look for <u>organic molecules</u>
- Definition of organic molecules
- contains C and H
- Examples
 - <u>sugars,</u>
 - <u>fats,</u>
 - proteins,
 - <u>nucleic acids</u>

Curiosity Rover landed in 2012



Curiosity lands on Gale Crater 2012

- 1. Evidence of an ancient streambed
- 2. Found CNHOPS
- 3. Found organic compounds in rocks
- 4. Evidence of methane production

Was there life on Mars

?



Chemical Bonds (Hold atoms together)

• Chemical bonds contain energy

• What happens when you break chemical bonds?

• What do you have to add to make big compounds?

3 types of chemical bonds

• Ionic = weak (hold salt together)

• Covalent = strong (holds organic molecules together)

- <u>Hydrogen bonds</u> = temporary
 - <u>hold water molecules together</u>



Compounds

- 2 or more atoms chemically bound together
- 2 types
 - Organic = C and H (ex: sugars, fats, proteins, nucleic acids)
 - -<u>Inorganic (ex: H₂O, O₂, CO₂)</u>
 - <u>Living things are made up of both</u>





4 types of Organic Compounds necessary for life

- Carbohydrates (sugars)
- <u>Proteins</u>
- Lipids (fats)
- Nucleic acids (DNA and RNA)

Organic compounds = stored energy

- Carbon containing molecules have a lot of chemical bonds
- <u>Chemical bonds = energy</u>
- Break down organic compounds → release energy
 - Ex: <u>Combustion \rightarrow releases energy</u>
 - Ex: Chemical digestion and respiration \rightarrow release energy

Post-assessment Quiz

- What are living things made of
 - Organic molecules
 - Inorganic molecules
 - Both Inorganic and inorganic molecules
- •
- List the 4 types of organic molecules
- What is the difference between organic and inorganic molecules
- What happens when you break chemical bonds

Go over Regents Practice Questions

Raw materials = $\underline{Nutrients}$

• Living things are made up of CHNOPS = 6 most important elements of life

• These nutrients must recycled → readily available to support life

• ROCK CYCLE AND PLATE TECTONICS IMPORTANT – Earth = only planet with plate tectonics

Decomposers

- Definition
- Organisms that recycle nutrients by consuming dead organisms
- Ex:
- Bacteria and fungi

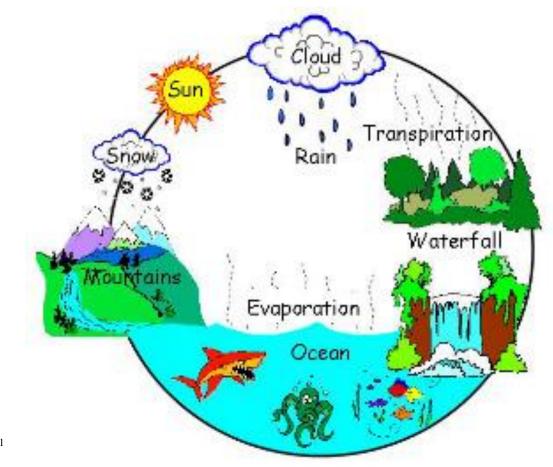


Water and Nutrient cycling

Why is Earth the only planet filled with life?

- <u>Liquid water</u>
- <u>Atmosphere</u> → right amount of solar energy (not too much and not too little)
- Plate tectonics \rightarrow <u>nutrient cycling</u>

Water cycle activity in your notes



http://water.tamu.edu/watercycle.html

Things that put water into the atmosphere

- Evaporation = $\underline{sun} \rightarrow \underline{liquid water} \rightarrow \underline{water vapor}$ (gas)
- Sublimation = $\underline{Sun} \rightarrow ice \rightarrow water vapor (gas)$

- Transpiration = $\underline{sun} \rightarrow \underline{leaves} \rightarrow \underline{water \ leaves}$ <u>trees as a gas</u>
- Respiration = <u>living things breath out CO_2 and <u>H₂O</u></u>

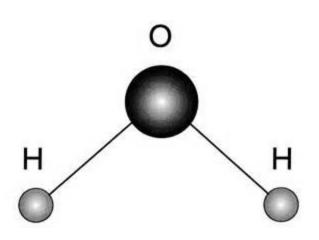
Processes that bring water down to the surface of the earth

• Condensation = water vapor (gas) \rightarrow liquid water

• Precipitation = <u>water falls from the sky</u> (rain, snow, sleet...)

• Runoff = <u>water flows on land into lakes</u>, <u>rivers and streams</u>

Draw a picture of a water molecule and state why it is so important to life



Draw a picture of the water cycle using all of the words from todays lesson

- Evaporation
- Transpiration
 - Sublimation
- Condensation
- Precipitation
 - Respiration
 - Runoff



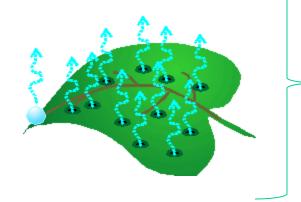
NOTES

Water cycle summary

Water \rightarrow atmosphere Evaporation = sun \rightarrow water vapor (gas)

- Transpiration = sun \rightarrow water leaves plants as a gas
 - <u>Stomates</u> = <u>openings in leaves</u>
 - Guard cells control size of opening
 - \rightarrow <u>control water loss</u>
 - Maintain homeostasis





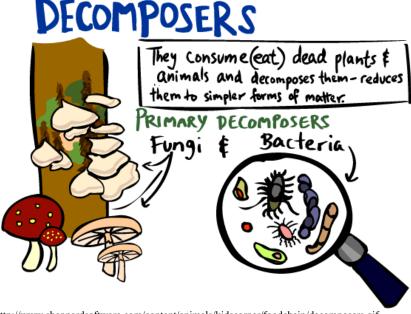
Example of Feedback Mechanisms

Regents Practice Questions

Elements must be recycled in

ecosystems

- Decomposers recycle nutrients (C, N, P, S...)
 - Bacteria and Fungi
 - Release elements trapped in living tissues
 - <u>Plants use the inorganic nutrients released by</u> <u>decomposers</u>

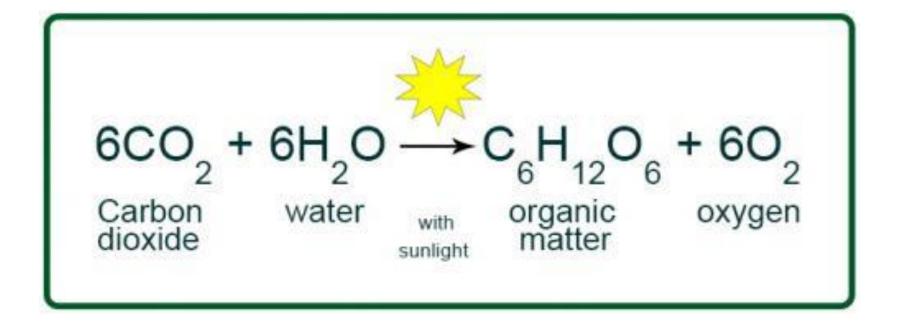


The Carbon Cycle

• 2 processes affecting the carbon cycle = Photosynthesis and Respiration

• Write the formula for each in your notes

Formula for photosynthesis (process that happens in plants)



Formula for respiration (process plants and animals use to release energy)

Cell Respiration Formula

C6H12O6 + 6O2 ->6CO2 + 6H2O + ATP

Glucose Oxygen Carbon Water Energy Dioxide



Carbon Cycle

- Carbon cycles between air, land and organisms
- All living things contain carbon
- **Fossil fuels** (coal, oil, gas)
 - come from <u>remains of living organisms</u>
 - All fossil fuels contain carbon
- **Photosynthesis** $\rightarrow \downarrow CO_2$ in air

Processes that release C into the air

- <u>Aerobic Respiration organism use $O_2 \rightarrow \uparrow CO_2$ in air</u>
- <u>Anaerobic respiration organisms</u> without oxygen $\rightarrow \uparrow$ methane in air (CH₄)
- **Decomposition** $\rightarrow \uparrow CO_2$ in air
- <u>Combustion (burning)</u> $\rightarrow \uparrow CO_2$ in air
- Volcanoes release carbon

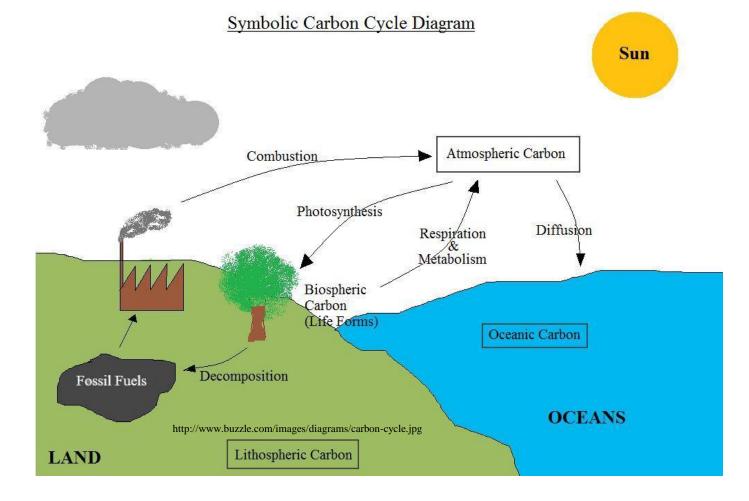
Note only 2 ways to take carbon out of air

- Photosynthesis = process that pulls carbon out of the air.
- Therefore <u>plants and forests</u> = <u>carbon sinks</u>

- Air mixes with the ocean
- Therefore <u>the ocean = carbon sink</u>
- when CO₂ enters the ocean it forms carbonic acid

Draw a picture of the carbon NOTES cycle

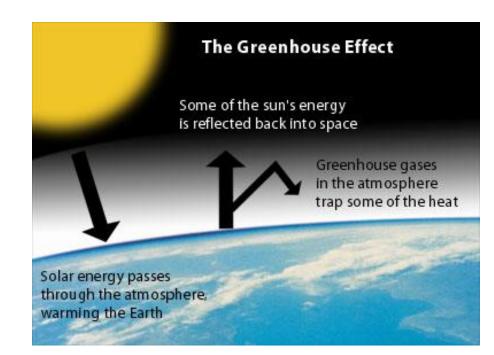
0(



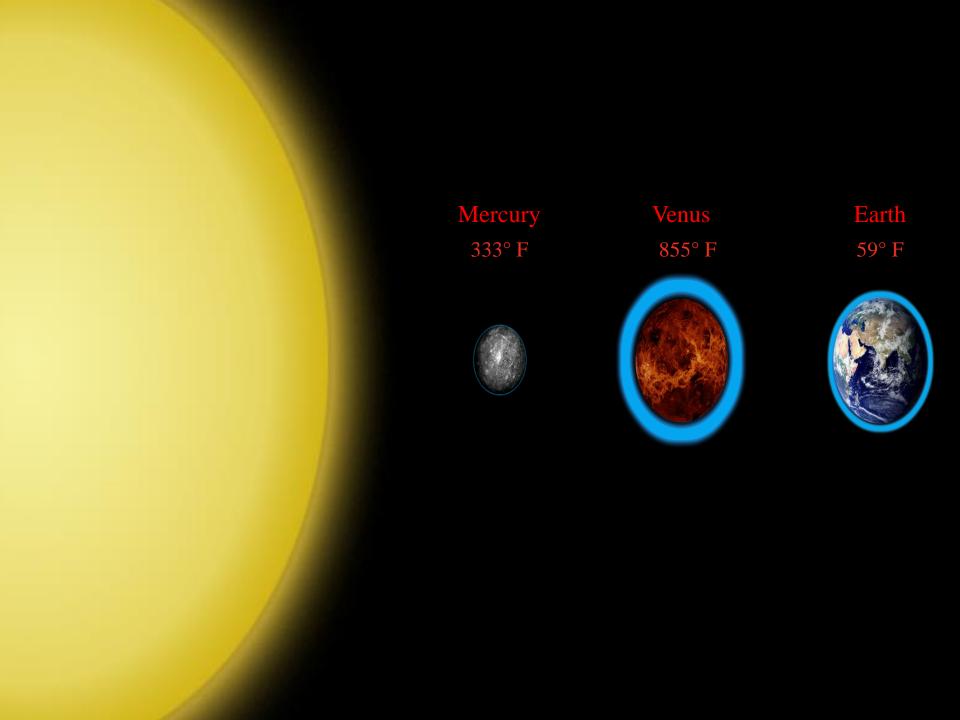
Human impacts on the Carbon cycle

$CO_2 CH_4, H_2O, NO_x = greenhouse$ gases

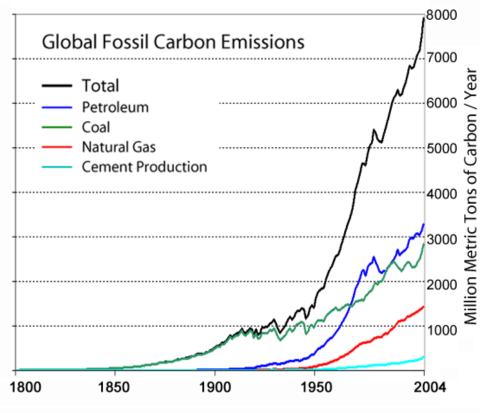
• Gases that trap heat near surface of Earth



IS THE GREENHOUSE EFFECT GOOD OR BAD???



Atmospheric CO₂

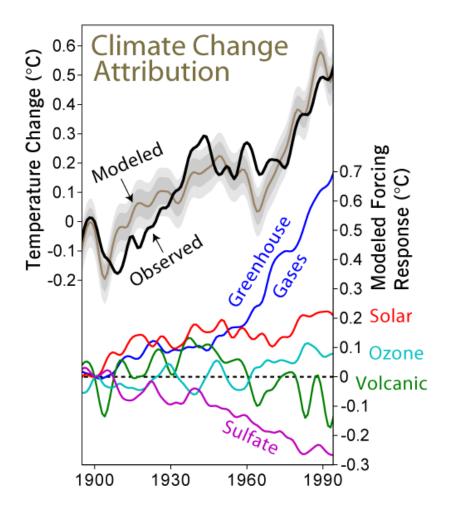


Data analysis questions

There are 2 time periods when the emissions rose

- When did atmospheric carbon emissions first begin to rise?
- 1850
- When did the rate of carbon emissions go up steeply
- 1950
- What was happening in the world during those times in history that might have caused the increase?
- 1850s = industrial revolution burning coal
- $1950s = oil \rightarrow gasoline \text{ for cars } \rightarrow \text{ increased transportation}$

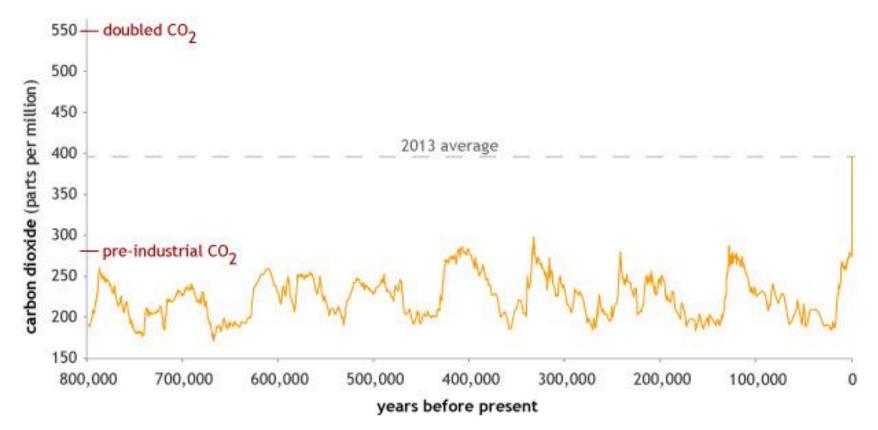
Climate Forcings (things that impact climates)



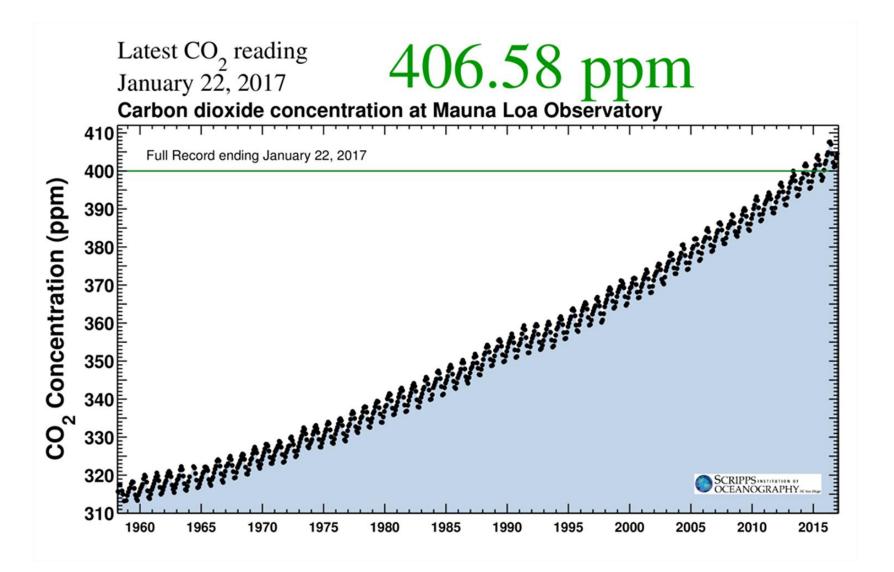
- How much as climate changed from 1900 1990?
- Increased about 0.5°C
- What do sulfates and volcanos do to weather
- Decrease temperatures
- What do GHGs do to climate?
- Increase temperatures

Atmospheric CO₂ levels

- Pre-industrial =
- <u>280 ppm</u>



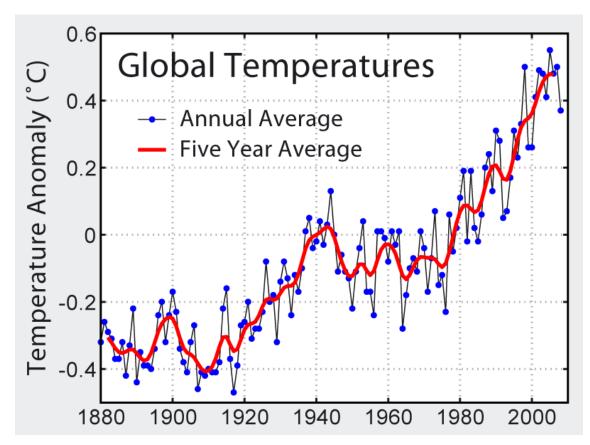
Atmospheric CO₂ levels today



Why?????

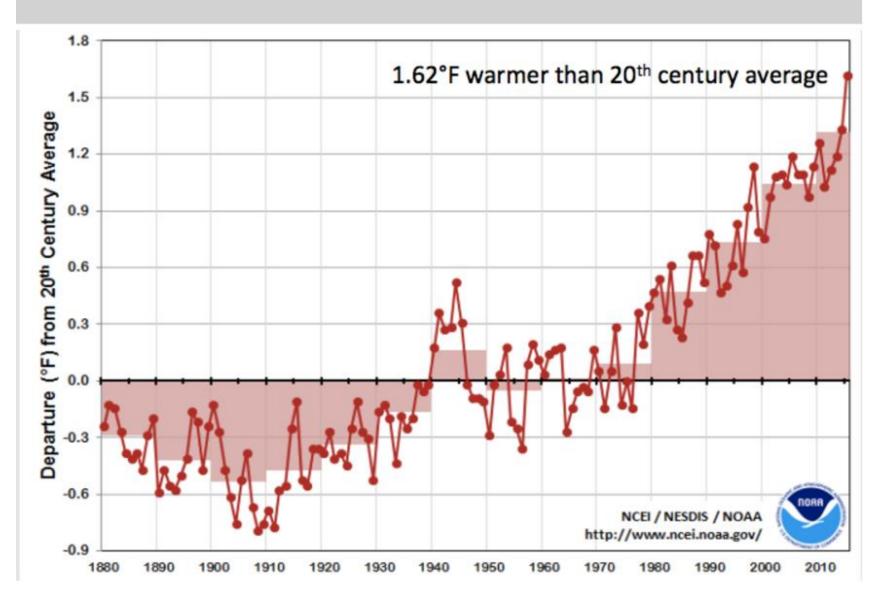
- Burning fossil fuels
- Deforestation

2007 IPCC Data indicates a <u>0.74°C rise in global</u> averages over the past century



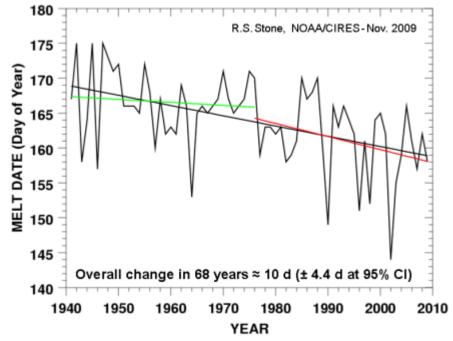
- This image shows the instrumental record of global average <u>temperatures</u> as compiled by the <u>NASA</u>'s <u>Goddard Institute for Space Studies</u>. The <u>data set</u> used follows the methodology outlined by Hansen et al. (2006). Following the common practice of the <u>Intergovernmental Panel on Climate Change</u>, the zero on this figure is the mean temperature from 1961-1990.
- Image created by Robert A. Rohde.

Global Temperature Time Series



Earlier spring snow melt in Barrow, Alaska

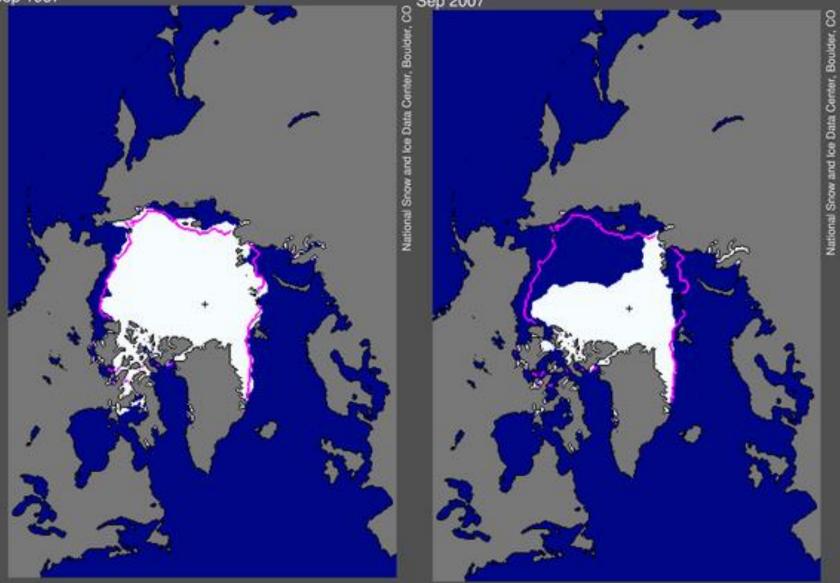
NOAA BRW melt date time series, 1941-2009



http://www.esrl.noaa.gov/gmd/grad /snomelt.html

Shrinking of polar ice sheets

Sea Ice Extent Sep 1987

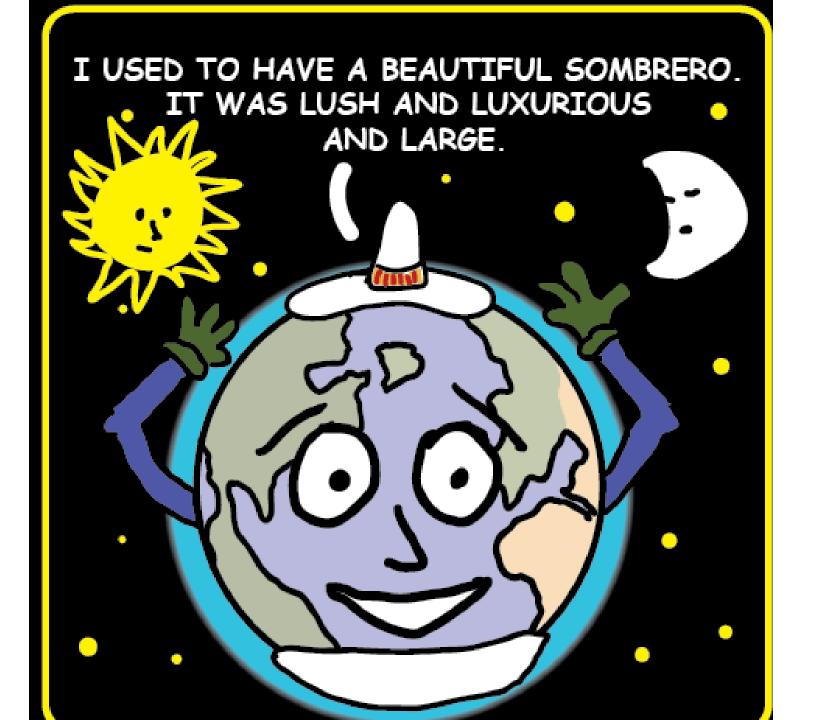


Total extent = 7.5 million sq km

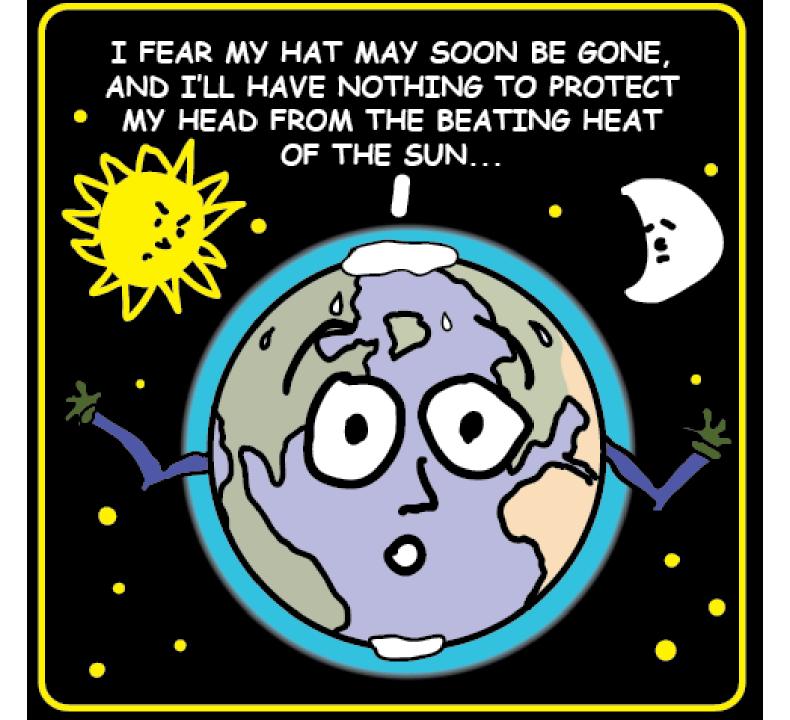
Total extent = 4.3 million sq km

median ice edge If the total sea ice extent in 1987 was 7.5 million square km and the 2007 value was 4.3 million square km, how much sea ice has been lost within this time period? (show your math)

- 7.5 million = 7.5×10^6
- 4.3 million = 4.3×10^6
- Difference = $3.2 \times 10^6 \text{ km}^2$











Why?????

- Burning fossil fuels
- Deforestation

Humans have a greater impact on ecosystems than any other organism due to their <u>ability to alter their</u> environment

